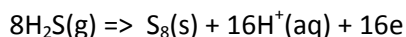
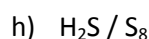
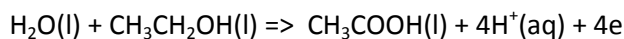
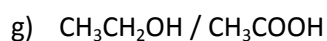
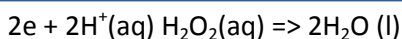
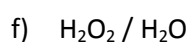
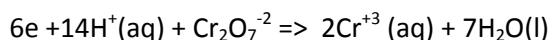
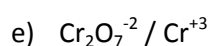
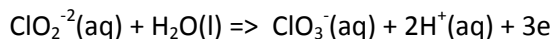
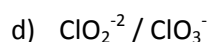
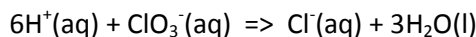
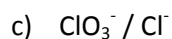
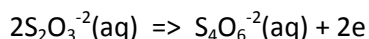
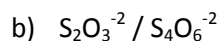
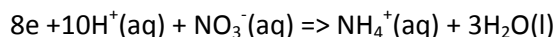
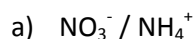
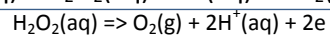
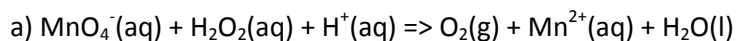


Revision/solutions

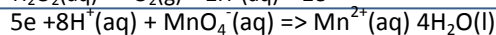
1) Write the balanced half equations to the following



2) Below are the unbalanced redox reactions. Give the balanced oxidation and reduction half reactions for each and the balanced overall equation.

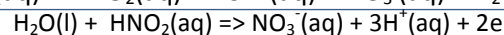
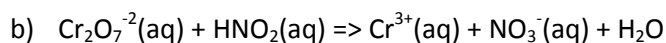


oxidation

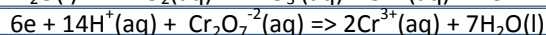


reduction

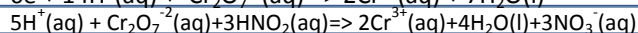
overall



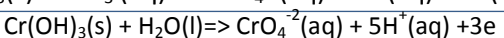
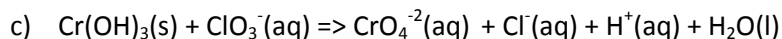
oxidation



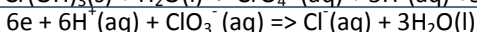
reduction



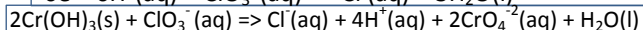
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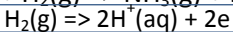
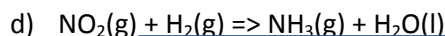
oxidation



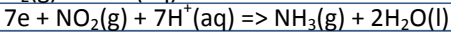
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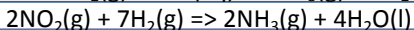
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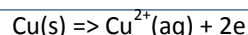
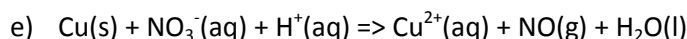
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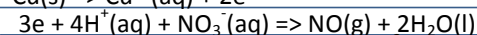
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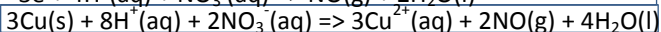
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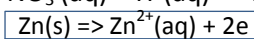
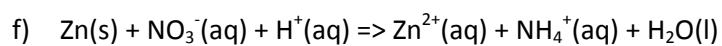
oxidation



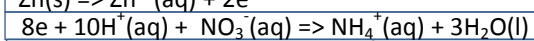
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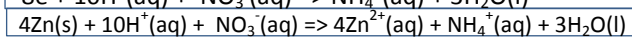
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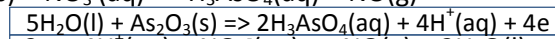
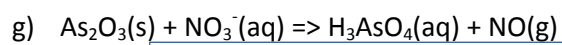
oxidation



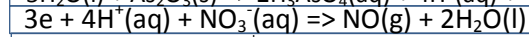
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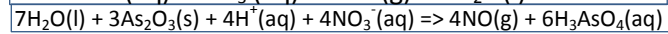
overall



oxidation



reduction



overall