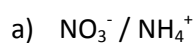
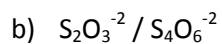
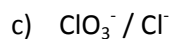


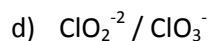
# Revision of half equations

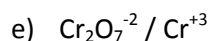
1) Write the balanced half equations to the following

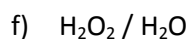


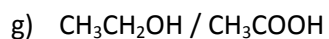


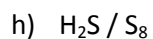




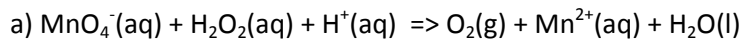




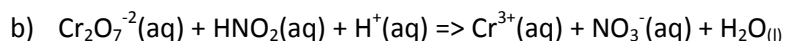




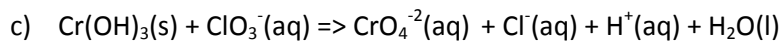

2) Below are the unbalanced redox reactions. Give the balanced oxidation and reduction half reactions for each and the balanced overall equation.

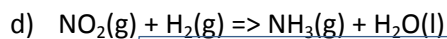

oxidation  
reduction  
overall

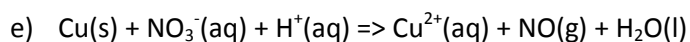

oxidation  
reduction  
overall

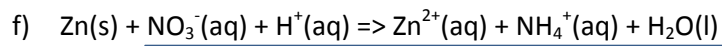

oxidation  
reduction  
overall

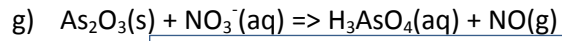

oxidation  
reduction  
overall

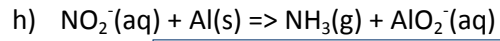

oxidation  
reduction  
overall



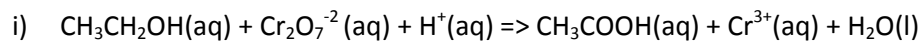

oxidation  
reduction  
overall



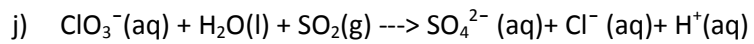

oxidation  
reduction  
overall



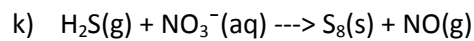

oxidation  
reduction  
overall



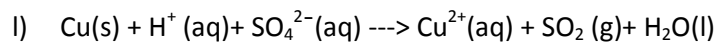

oxidation  
reduction  
overall




oxidation  
reduction  
overall




oxidation  
reduction  
overall




oxidation  
reduction  
overall