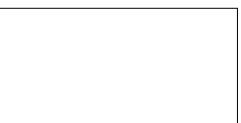
Organic worksheet 3

1) Give the systematic IUPAC name for the molecules shown on the right .

- A)
- B CH₃ CH₃ H₃C-CH-CH₂-C-CH₃ CH₃
- B) _____
- $\begin{array}{cccc} \textbf{C} & & & & \text{CH}_3 \\ \textbf{C} & & & & & \\ \textbf{H}_3\textbf{C} \textbf{C}\textbf{H} \textbf{C}\textbf{H}_2 \textbf{C}\textbf{H}_2 \textbf{C} \textbf{H} \\ & & & & \\ \textbf{C}\textbf{H}_3 \end{array}$
- C) _____
- 2) Give the structural and semi-structural formulae of the following molecules.
 - a) 3-ethylhexane



- b) 2, 4-dimethylpentane



- 3) Which one of the following compounds is most soluble in water at room temperature? Explain
 - a) Propane
 - b) Propene
 - c) Butan-2-ol
 - d) Butanoic acid
- 4) 3-bromobut-2-ene and HCl react according to the equation below.

- a) Give the possible structural formulae and systematic names of X
- b) What type of reaction is this?
- 5) Which of the following compounds are isomers of 2,2,4-trimethylpentane
 - a) octane
 - b) 3-ethylhexane
 - c) 2, 4-dimethylpentane
 - d) 2,4-dimethylhexane
- 6) Below is the diagram of the formation of halothane.

- a) Give the IUPAC systematic name for halothane
- b) What type of reaction is step 2?