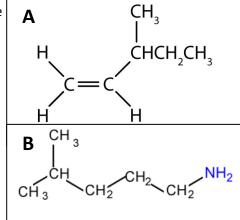
Friday Worksheet

Organic worksheet 2

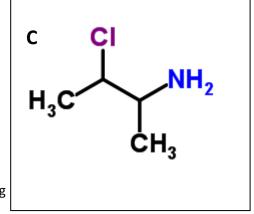
1) Give the systematic IUPAC name for the molecules shown on the right .

A) _____

R)



C) _____

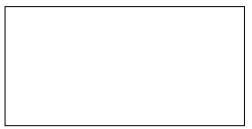


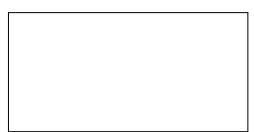
2) Give the structural and semi-structural formulae of the following molecules.

a) 3-Chlorobutan-2-amine



b) 2,3-dimethyl-1,3-butadiene





3)	1.35 g of an alkene that contains two double bonds per molecule reacted completely with 8.0 g of bromine, Br_2 . The molar mass of bromine, Br_2 , is 160 g mol ⁻¹ .
	a) Which one of the following is the molecular formula of the alkene? Explain how you arrived at your choice. Show all working out.
	1) C ₃ H ₄ ,
	2) C ₄ H ₆ ,
	3) C ₅ H ₈ ,
	4) C ₆ H ₁₀
	b) Draw structural formulae and name each isomer of the alkene.