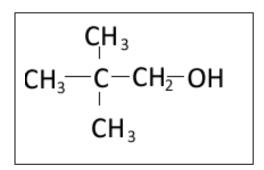
Friday Worksheet HNMR spectroscopy 3

Name:

1) Consider the molecule whose structural formula is shown below.

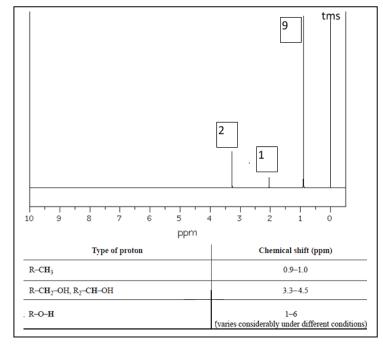


- a) Write the IUPAC name for the molecule. *2,2-dimethylpropan-1-ol*
- b) In the box, on the right, draw the
 ¹HNMR spectrum of this molecule.

 Note that the hydrogen on the OH group creates a signal at 2.0 ppm.

 The data sheet should be used to

determine the ppm of each non-equivalent group of hydrogens.

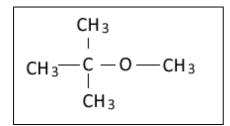


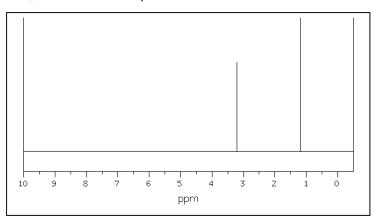
d) How many signals are expected on the ¹³CNMR spectrum? 3

c) Indicate on the spectrum the relative area of each signal.

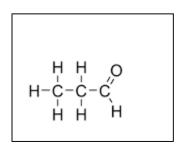
2) A compound has the molecular formula C₅H₁₂O. Its ¹HNMR spectrum is shown below.

If the ¹³CNMR spectrum shows only three signals draw the structural formula of this compound in the box below.

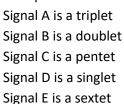




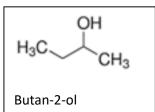
- 3) On the right is the ¹HNMR spectrum of an organic compound(X) with the molecular formula C₃H₆O. Three signals are visible, with two triplets at 9.5 and 0.8 ppm. A multiple peak signal at 3.0 ppm is also seen.
 - a) Draw the structural formula in the box below.

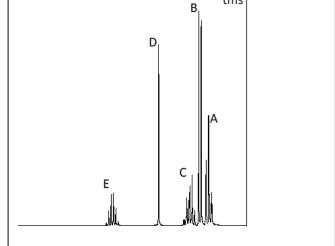


- b) Name and draw the structural formula of the compound formed when compound "X" is oxidised in the presence of acidified Cr₂O₇-2 solution.
- c) Another compound (Y) was analysed using ¹HNMR and its spectrum is shown below. The spectrum shows 5 signals and are labelled.



a) Name and draw the structural formula of compound Y if its molecular formula is C₄H₁₀O.





Propanoic acid

- b) Draw the structural formula of the compound formed when compound"Y" is oxidised in the presence of acidified $Cr_2O_7^{-2}$ solution.
- c) To what group of compounds does this product belong? *Ketones*.
- d) What is the functional group of this class of compounds?
- e) To what group of compounds does "Y" belong? secondary alcohol

