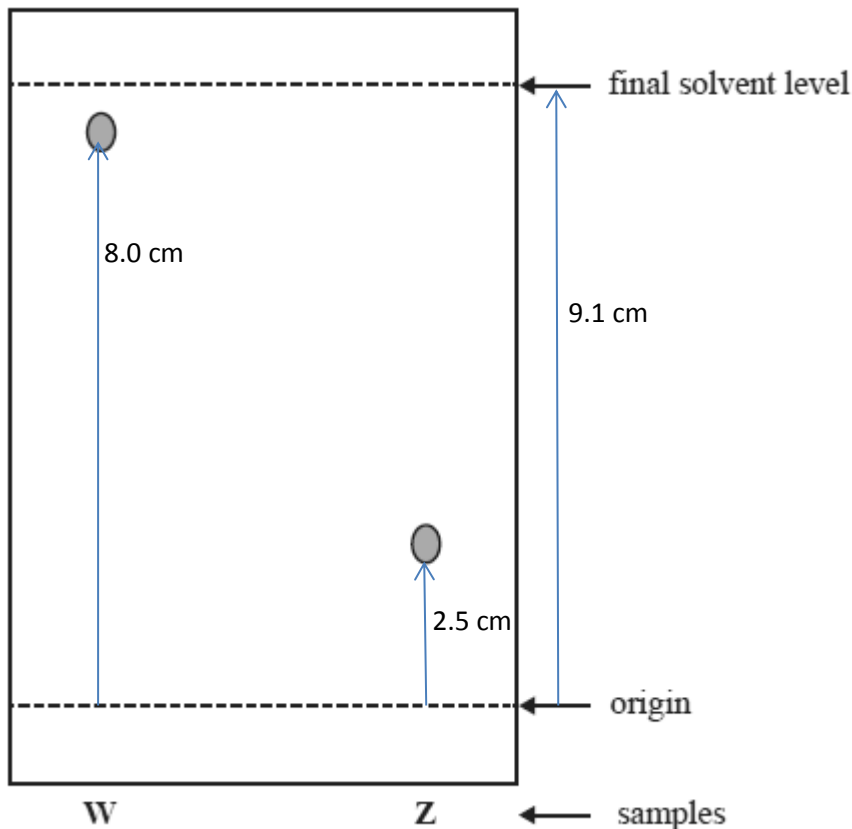


Chromatography 1

- 1) Two different food dye samples, W and Z were compared using thin layer chromatography as shown below.



- a) Calculate the R_f value of W and Z
 $W = 8.0/9.1 = 0.88$, $Z = 2.5/9.1 = 0.28$
- b) Which dye is more strongly adsorbed?
Z
- c) Explain how dyes W and Z can be identified?
Under these conditions they have a unique R_f value which can be used to identify these compounds
- d) On the right is another chromatogram of a food substance containing food dye. Is dye "W" present? Explain.
Yes
C has an R_f value of 0.88 same as W. If each chromatogram was run under identical conditions then the R_f value is identical to W.

